

Electrospun Nitrogen-doped TiO₂ Nanofibrous Thin Film for Photovoltaic Application



Mohamed Hasaan Hussain, Noor Fitrah Abu Bakar, Lim Ying Pei, Lim Ying Chin, Mohd Nazli Naim

Abstract: This research study aims at fabrication of fine size nitrogen doped TiO₂ nanofiber using electrospinning method and evaluation of the performance of TiO₂ in a photovoltaic cell under visible light irradiation. Undoped and N doped TiO₂ nanoparticles were synthesized by sol gel method where titanium isopropoxide was used as the source of TiO₂ and ammonium nitrate was used as the source of N dopant. TiO₂/PVA composite material was prepared by stabilizing TiO₂ particle in to 10 wt % of PVA (aq) solution in order to prepare thin film that can be coated on photovoltaic (PV) cells. Coating of solid thin film PV cells by TiO₂/PVA nanofibers was conducted using electrospinning and doctor blade method. In both systems, doping the TiO₂ with nitrogen improved its optical properties which it successfully lowered the band gap energy from 3.14 to 2.76 eV and shifted its optical response to the visible light region. The presence of O-H stretching vibration, O-H bending and vibration of the N-Ti bond contributed to an increased performance of the PV cells. The electrospun N-doped TiO₂ produced better power output than doctor blade method coated PV cells with power of 0.040 and 0.026 mW, respectively.

Index Terms: Electrospinning, Nitrogen doping, Photovoltaic, TiO₂.

I. INTRODUCTION

Semiconducting nanoparticles have emerged as an inevitable component in common energy sources due to their optical and catalytic properties, and it is enhancing alternative energy method to help on the world's rising energy demands [1]. Numerous researchers are looking forward in developing affordable, renewable and clean energy sources, and to reduce energy consumption as well as toxicity release to the environment with the aid of these nanoparticles [2]. Notably,

TiO₂ nanoparticles have been intensively studied for its ability to photocatalytic and applications have high performance for photovoltaic application at cheap cost. It also shows low pollutant load, low toxicity, chemically stable and available at low price [3]. Although TiO₂ is recognized as an efficient semiconductor and photo catalytic material, the high energy band gap of TiO₂ limits its solar light absorbance capability and photovoltaic current generation [4],[5]. In order to overcome this issue by enhancing the performance of TiO₂ to absorb visible region solar light, several techniques such as sensitizing TiO₂ with dye and doping the TiO₂ with metal and non-metal element have been proposed [5], [6], [7], [8]. Among all these methods, doping TiO₂ by nitrogen (N) atoms has been studied intensively and N modification in TiO₂ is believed to narrow down the energy band gap efficiently and enlarge the solar light wavelength range [9]–[12].

Beside improving the chemical properties, increasing the surface area of TiO₂ film layer improve the photocatalytic efficiency by absorbing high amount of sensitized dye that lead high amount of light harvesting [13]. Also, high surface area of TiO₂ thin film allow electrolyte to diffuse through the cell intensively and results high photo electrochemical reaction rate. In addition, one dimensional nanostructure of TiO₂ thin film such as nanofibers increase the electron diffusion co efficient and results energy production to a great extent [14]. Considering these facts, TiO₂ thin film with highest surface area and high porosity can be constructed using TiO₂ – polymer composite nanofibers in order to increase the efficiency in photovoltaic applications. In this regard, electrospinning method has been proposed as a versatile method to synthesize metal nanoparticles incorporated nanofibers for various applications from biomedical field to energy applications.

Electrospinning is an advanced technique that produces continuous nanofibers from polymer solutions using electrical forces [15], [16]. By using this method, the physical properties of the polymers such as diameter, porosity, interconnectivity, orientation and compositions can be manipulated, and super thin films can be constructed with optimum parameters [16]–[18]. It is expected that these characteristics of electrospun thin films can aid to enhance the efficiency of energy generation. For instance, many studies have reported that metal nanoparticles incorporated electrospun fiber can be used as highly sensitive electrodes [19], [20], proton exchange membrane [21] and catalytic layers [22].

Manuscript published on November 30, 2019.

* Correspondence Author

Mohamed Hasaan Hussain*, Faculty of Chemical Engineering, Universiti Teknologi MARA, 40450, Shah Alam, Malaysia.

Noor Fitrah Abu Bakar*, corresponding author, Faculty of Chemical Engineering, Universiti Teknologi MARA, 40450, Shah Alam, Malaysia.

Lim Ying Pei, Faculty of Chemical Engineering, Universiti Teknologi MARA, 40450, Shah Alam, Malaysia.

Lim Ying Chin, Faculty of Applied Sciences, Universiti Teknologi MARA, 40450, Shah Alam, Malaysia.

Mohd Nazli Naim, Faculty of Engineering, Universiti Putra Malaysia, Selangor, 43400 Serdang, Malaysia.

© The Authors. Published by Blue Eyes Intelligence Engineering and Sciences Publication (BEIESP). This is an [open access](http://creativecommons.org/licenses/by-nc-nd/4.0/) article under the CC-BY-NC-ND license (<http://creativecommons.org/licenses/by-nc-nd/4.0/>)

In this work, the electrospinning technique is used to produce TiO₂ nanofibers with a high surface area that is capable to harvest a large amount of light and act as a sensitive electrode for PV cells. N doped TiO₂ and undoped TiO₂ incorporated PVA composite nanofiber thin film was produced using electrospinning method and highly efficient. Plant based natural dyes are recommended as an efficient and environmentally friendly source for absorbing visible dyes [23]. PV cell was constructed utilizing this modified TiO₂/PVA composite film. As a comparison, doctor blade method was used for coating and preparing of the PV cell.

II. MATERIALS AND METHOD

A. Synthesis Undoped and N-Doped TiO₂ Nanoparticles

First, N-doped and undoped TiO₂ nanoparticles were synthesized using sol-gel method. Titanium (IV) isopropoxide (C₁₂H₂₀O₄Ti, Sigma Aldrich) was used as the source of the TiO₂ and ammonium nitrate (NH₄NO₃, Sigma Aldrich) was chosen as the nitrogen source for N doped-TiO₂. During the sol-gel preparation, 6 ml of concentrated nitric acid (HNO₃) was added to 200 ml of distilled water. To prepare 10 wt% N doped TiO₂ suspension, 3.75g of NH₄NO₃ was added into the 40 ml of titanium (IV) isopropoxide and the resultant mixture was continuously stirred at 500 rpm for 1 hour at room temperature. For preparing undoped TiO₂ suspension, the same procedure was repeated without NH₄NO₃ dopant. Secondly, in order to prepare dry N doped TiO₂ and TiO₂ nanoparticles, obtained TiO₂ suspensions were poured into the crucible for drying and calcining. For the drying process, the suspension was dried at 90°C for 3 hours in an oven. After that, the dried materials were calcined at 400°C for 2 hours in air. The white crystal which consists of TiO₂ was obtained and the materials were then further ground into a powder by using pestle and mortar. White undoped and 10wt% N-doped TiO₂ powder was obtained.

B. Undoped and N-doped TiO₂ /PVA Polymer Solution

In order to prepare composite precursor solution, typical 10 wt% of polyvinyl alcohol (PVA) (M_w150000) (Sigma Aldrich) solution was prepared by dissolving 2g of PVA powder into 20 ml distilled water under vigorous stirring for 5 hours. Subsequently, 1 g of 10wt% N-doped TiO₂ powders were added into the 20 ml, 10 wt% of PVA aqueous solution in separate vials and all resultant mixture was continuously stirred for 30 mins.

C. Electrospinning of Undoped and N-Doped TiO₂/PVA Blended Solution

Electrospinning apparatus was set up by using 5 ml disposable plastic syringe equipped with a 23 G flat-end metal needle, syringe pump and high voltage power supply as depicted in Fig. 1. The blended solution TiO₂ – PVA precursor solution was transferred into the 5 ml disposable plastic syringe without air bubbles trapped into the syringe. At the counter electrode, an indium tin oxide coated (ITO) glass with a dimension of 2cm x 2cm glass was placed at the middle of the collector plate. The conductive side of the ITO glass was set to face the syringe pump in order to deposit electrospun fiber on the surface. The conductive side of the

ITO glass can be tested by using a multimeter, where it should be continuity at the conductive glass surface. The counter electrode of the electrospinning set was earthed and the distance between the collector plate and the needle tips was fixed at 15cm. During the electrospinning process, 15 kV high voltage was applied to the metal capillary (needle) using a high voltage power supply. The syringe pump was set to feed the precursor solution with the constant flowrate of 2mL/h. Eventually, undoped and N-doped TiO₂ nanofiber were deposited on the ITO glass and electrospun fiber-coated glass was calcinated at 400°C for 1 hour. After the calcination process, the ITO glass that was deposited with TiO₂ nanofiber were cooled down to the room temperature.

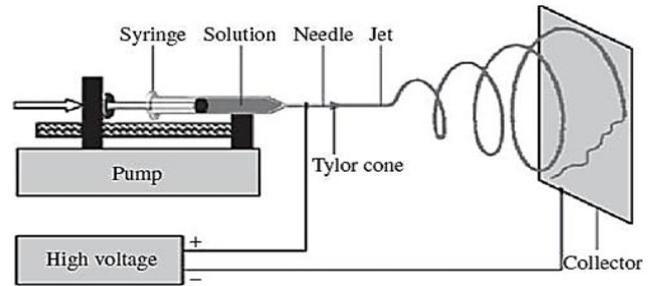


Fig. 1. Schematic diagram of electrospinning experimental setup for undoped and N-doped TiO₂/PVA fiber production

D. Doctor Blade Method Deposition of Undoped and N-Doped TiO₂/PVA Blended Solution

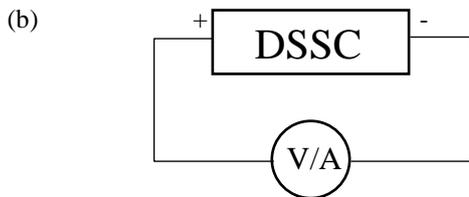
First, the ITO glass was taped at both the left and right side of the ITO glasses and undoped TiO₂/PVA blended paste was applied at the top edge of the ITO glass between the two pieces of tape. Then the blended paste was spread across the ITO glass using a glass rod. The gap between the strips of tape was filled with a layer of the blended solution. The same procedure was repeated with the N-doped TiO₂/PVA blended solution as well. After the undoped and N-doped TiO₂/PVA blended were deposited on the ITO glass by doctor blade method, the ITO glasses were calcinated at 400°C using a furnace. After the calcination process, the ITO glass was cooled to the room temperature.

E. Staining of TiO₂ /PVA Thin Film and Photovoltaic Cell Assembly

Plant based natural dyes are recommended as an efficient and environmentally friendly source for absorbing visible dyes [23]. Anthocyanin pigment dye that was extracted from the flesh dragon fruit, was used for staining and sensitizing TiO₂ composite thin film in this study. Firstly, flesh dragon fruit was crushed to obtain juice and placed into the petri dish. The deposited surface of ITO glasses of undoped and N-doped TiO₂ /PVA were immersed into the dye for 10 minutes. During the soaking process, dye molecules were absorbed by the TiO₂ particles and the coated surface of ITO glasses was stained.

Photoelectrochemical cell preparation and photoelectrochemical measurements were carried out as shown in Fig. 2. A carbon counter electrode was prepared by creating and coating the ITO glass with carbon soot under a candle. Afterward, dye sensitized TiO₂ coated electrode and carbon counter electrode were clamped together using paper clips. As a redox electrolyte solution, a small amount of anhydrous ethylene glycol dissolved potassium iodide/iodine solution was poured between the sandwiched electrode. Fig. 2 shows the diagram of the full assembly dye sensitized solar cell. The photovoltaic was then analyzed by a digital multimeter to obtain the data for PV cell performance.

A total of four samples of dye sensitized TiO₂ solar cells produced which consist of two methods, doctor blade method and electrospinning method. The performance of the photovoltaic cells produced in this experiment was determined by measuring the open circuit voltage, V_{oc}, and short circuit current density, J_{sc} by using digital multimeter. The photovoltaic cell was tested under the sunlight at three different times which was at 1.00 PM, 3.00 PM and 5.00 PM.



V = Voltmeter
A = Ammeter
DSSC = Dye Sensitize Solar Cell

Fig. 2. (a) Dye sensitized TiO₂ coated ITO glass electrodes, (b) dye sensitized TiO₂ PV cells for voltage and current measurements.

F. Characterization of the TiO₂/PVP Composite.

Morphologies of the electrospun TiO₂/PVP composite fibers were examined by Field Emission Scanning Electron Microscope (FE-SEM; Zeiss Supra 40). The samples were sputtered with gold.

UV-Visible Spectroscopy (UV-Vis; Agilent Cary 60 spectrometer) was used to study the changes in absorption wavelength for both undoped and N-doped TiO₂ samples. After electrospinning, the TiO₂/PVP composite fibers were peeled from the aluminum foil and characterized. UV-Vis

absorbance spectra of all samples were recorded at a wavelength range from 200 nm to 800 nm.

Fourier Transform Infrared Spectroscopy (FTIR; Perkin Elmer) was used to study the changes in the functional group for the electrospun N-doped TiO₂ samples. After electrospinning, the deposited fibers were removed from the aluminum foils and tested. The scan spectra were recorded in the range from 500 cm⁻¹ to 4000 cm⁻¹.

III. RESULTS AND DISCUSSION

A. Chemical Bonding and the Chemical State of Undoped and N-Doped TiO₂

The bonding characteristics of functional groups in N-doped TiO₂ samples and undoped TiO₂ sample were identified by FTIR. Fig. 3 shows the absorption peaks from FTIR spectra for the N-doped TiO₂ samples at two different concentrations, undoped sample and samples after the calcination process. The observed peak in the range of 3324.94 cm⁻¹ -3369.00 cm⁻¹ is attributed to O-H stretching vibration and the peak detected at 1634.70 cm⁻¹ -1635.06 cm⁻¹ is assigned to O-H bending. It is known that hydroxyl group plays an important role in the photocatalytic mechanism. Hydroxyl group was trapped by hole generated when TiO₂ was irradiated under UV light. Hydroxyl radicals formed suppress the recombination of electron-hole formation, hence increasing the TiO₂ photocatalytic efficiency [24]. The peaks around 1433.73 cm⁻¹ and 1438.12 cm⁻¹ are attributed to the vibration of the N-Ti bond [25]. There were few absorption peaks in the vicinities of 1700 cm⁻¹ to 800 cm⁻¹ from chelating of PVA with the titanium compound. After calcination at 400°C, the PVA absorption peaks were disappeared, indicating the decomposition of PVA.

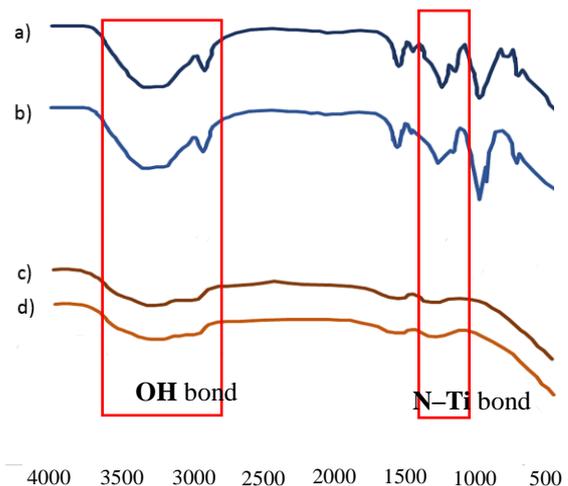


Fig. 3. FTIR spectra for (a) undoped titanium oxide, (b) N-doped titanium oxide, (d) undoped titanium oxide after calcination (c) N-doped titanium oxide after calcinations

B. Optical Properties and Band Gap Analysis of TiO₂ Thin Film Synthesized by Electrospinning Method

Undoped and N doped-TiO₂ thin films prepared by the electrospinning method were initially analyzed by UV-Vis technique in order to find out the absorption spectra and determine the band gap energy. The optical response for the undoped and N-doped TiO₂ samples were shown in Fig. 4. As seen in the Fig.4, the absorbance spectrum of N-doped TiO₂ sample significantly shift to the higher wavelength region and cut off wavelength of undoped TiO₂ is 395 nm whereas cutting off wavelength of N-doped TiO₂ increased to 450 nm. Absorption spectra shift to the higher wavelength of the visible region in the presence of dopant indicate that N-doped TiO₂ absorbs wider wavelength of light than the undoped TiO₂.

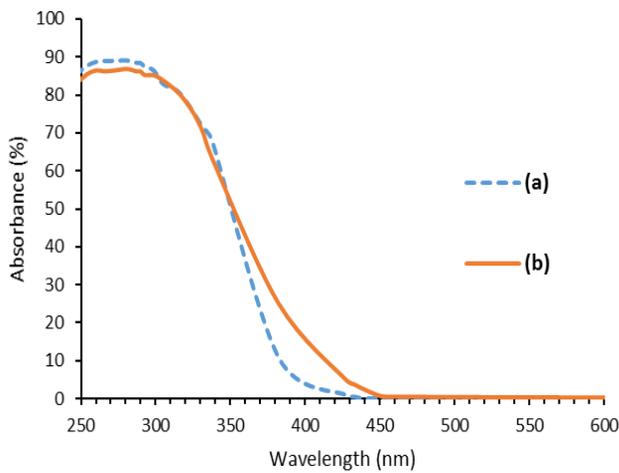


Fig. 4. UV-Vis absorption spectra of (a) undoped TiO₂ and (b) N-doped TiO₂

The relationship between the band gap energy and the cut off wavelength in absorption spectra was shown in equation (1) whereby band gap energy, E_{bg} (eV), plank constant, h=6.626 x 10⁻³⁴ Js and speed light, c=3.0 x 10⁸ m/s:

$$E_{bg} = \frac{hc}{\lambda} \tag{1}$$

The energy band gap of TiO₂ electrospun membranes was determined using equation (1). Determined band gap energy along with the corresponding cut off absorbance wavelength are shown in Table I. Obtained band gap energy of the undoped TiO₂ sample was 3.14 eV. This value is consistence with the band gap range of TiO₂ that has been reported in the literature as 3.0 eV - 3.2 eV.

Table I. Band gap energy for undoped and N-doped TiO₂ samples

Sample	Cut-off wavelength (nm)	E _{bg} (J)	E _{bg} (eV)
undoped TiO ₂	395	5.032 x 10 ⁻¹⁹	3.14
N-doped TiO ₂	450	4.417 x 10 ⁻¹⁹	2.76

When nitrogen dopants were introduced in TiO₂, the band gap energy of N -TiO₂ reduced to a lower level which was 2.76 eV. This result indicates that N doping considerably narrows down the band gap energy of TiO₂. When TiO₂ is doped with N, N atoms distort the TiO₂ lattice and occupy the interstitial or substitutional position of the lattice [26], [27]. In this position, N 2p state overlap with O 2p and significantly rise the localized state of the oxide near to the top of the valency band or create a new inter band site that aid to pass the energy gap [26].

C. Morphology Analysis of Electrospun Undoped and N-doped TiO₂ / PVA Polymer Composite Fiber

FESEM micrographs of electrospun N-doped TiO₂/PVA and undoped TiO₂/PVA composite nanofibrous are shown in Fig. 5. Corresponding images of N-doped TiO₂/PVA (Fig. 5(a) and (b)) and undoped TiO₂/PVA (Fig. 5(c) and (d)) prove that continuous, discrete and randomly distributed TiO₂/PVA composite nanofiber has been produced by electrospinning technique. Furthermore, randomly elongated lumps appear on N-doped TiO₂/PVA composite fiber surface. It is speculated that these lumps ascribe to the dispersion of TiO₂ nanoparticle and these results illustrate that TiO₂ nanoparticles were consequently dispersed in PVA nanofiber [28].

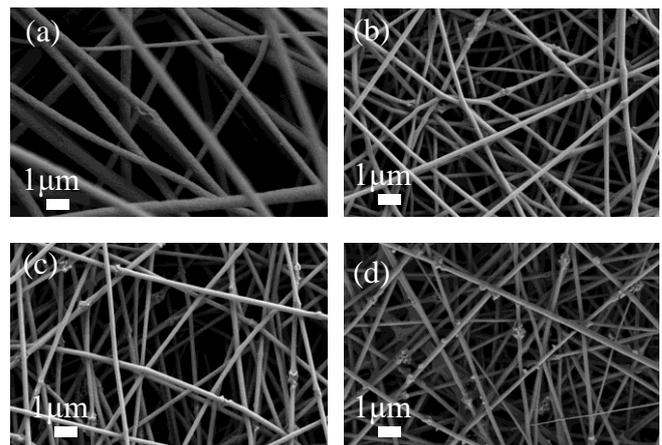


Fig. 5. FESEM images of undoped and N-doped TiO₂ electrospun fibers, (a) undoped TiO₂ fibers before calcination, (b) undoped TiO₂ fibers after calcination, (c) N-doped TiO₂ before calcination and (d) N-doped TiO₂ after calcination

Table 2 shows the mean diameter of undoped and N-doped TiO₂/PVA nanofibers. According to Table II, size of both N-doped TiO₂/PVA and undoped TiO₂/PVA were in the range of 200 nm to 700 nm which is the optimum size that offering a large surface area for the membrane [29]. However, the mean diameter of both calcined N-doped TiO₂/PVA and undoped TiO₂/PVA were 304±43 nm and 672±19 nm respectively whereas these values decreased to 202±72 nm and 342±38 nm after calcination. These changes in fiber diameter were obtained due to the evaporation of partial PVA and solvent molecules from the fiber during the calcination process [30],[31].

Therefore, it was expected that the higher rigid surface area and spaces between the fibers may harvest more solar energy and stimulate higher photovoltaic reaction than the continuous thin films.

Table II. Mean diameter of undoped and N doped TiO₂ fibers before and after calcination

Sample	Mean diameter before calcination	Mean diameter after calcination
Undoped TiO ₂	672±19 nm	342±38 nm
N-doped TiO ₂	304 ± 43 nm	202±72 nm

D. Photovoltaic Performance of Electrospun and Doctor Blade Method of Undoped and N-doped TiO₂ on Photovoltaic Application

Photovoltaic performances of doctor blade method of N-doped TiO₂ and undoped TiO₂ PV cells have been tabulated in Table III. Here in, produced circuit voltage (V_{oc}) and photo current density (J_{sc}) of each PV cells were recorded at different time period (1pm, 3pm and 5pm) of the same day in order to determine the average power output of doctor blade method and electrospun TiO₂ coated PV cells. Results show that the dye sensitized N-doped TiO₂ PV cells exhibit higher voltage and current at every time than the undoped TiO₂ PV cells. Overall, the average power produced by the N-doped TiO₂ PV cells (0.026 mW) was higher than the undoped TiO₂ PV cells (0.022 mW). These results are in agreement to the previous speculation which is N dopant has significantly increased the light absorbance efficiency of TiO₂

and enhanced photocatalytic performance of TiO₂ PV cells. Table IV shows the photo current density and voltage, generated by electrospun coated N-doped TiO₂ and undoped TiO₂ PV cells. Electrospun N-doped TiO₂ PV cells generated average 245.7 mV V_{oc} and 40.7 μAcm⁻² J_{sc} whereas electrospun undoped TiO₂ generated 178.7 mV V_{oc} and 34.9 μAcm⁻² J_{sc}. These observations comply with the previous finding that N-doped TiO₂ has efficiently absorbed wide range of light wavelengths and produced more energy than undoped TiO₂ [32],[33]. Table III and IV summarize that electrospun N-doped TiO₂ composite PV cells performed better than the doctor blade method. Generally, the higher value of V_{oc} and J_{sc} indicate the higher power output and better performance, which lead to a higher efficiency of photovoltaic cells [34]. In both N-doped TiO₂ and undoped samples electrospun coated PV cells produced 0.025 mW and 0.040 mW power respectfully whereas doctor blade method coated PV cells produced 0.022 mW and 0.026 mW power respectively.

Therefore, these results implicit that the electrospun TiO₂/PVA composite fiber has a great potential to produce photovoltaic current than TiO₂/PVA solid thin films. It is hypothesized that these higher power output of electrospun fibers obtained due to the high surface area of nanofibers that can harvest high amount of solar light and orientation of the fiber may improve the electrode mobility and charge transfer inside the cell [35]. In addition, the gap between the fibers (c.f Fig. 5) may trap the solar light efficiently tend to enhance the photovoltaic reaction.

Table III. Voltage and current density of doctor blade method coated titanium dioxide solar cells at different time interval

Time	1.00 PM		3.00 PM		5.00 PM		Average		Power
	V _{oc} (mV)	J _{sc} (μAcm ⁻²)	V _{oc} (mV)	J _{sc} (μAcm ⁻²)	V _{oc} (mV)	J _{sc} (μAcm ⁻²)	V _{oc} (mV)	J _{sc} (μAcm ⁻²)	
Undoped TiO ₂	158	35.7	153	35.6	149	35.6	153.3	35.6	0.022
N-doped TiO ₂	179	38.0	169	37.9	166	37.8	171.3	37.8	0.026

Table IV. Voltage and current density of electrospun coated titanium dioxide solar cells at different time intervals

Time	1.00 PM		3.00 PM		5.00 PM		Average		Power
	V _{oc} (mV)	J _{sc} (μAcm ⁻²)	V _{oc} (mV)	J _{sc} (μAcm ⁻²)	V _{oc} (mV)	J _{sc} (μAcm ⁻²)	V _{oc} (mV)	J _{sc} (μAcm ⁻²)	
Undoped TiO ₂	184	35.2	177	34.8	175	34.7	178.7	34.9	0.025
N-doped TiO ₂	250	41.3	248	40.6	239	40.3	245.7	40.7	0.040

IV. CONCLUSION

Undoped and N-doped TiO₂/PVA composite were deposited on ITO glasses by electrospun methods and doctor blade method. Initially, UV-Vis absorption spectra of undoped and N-doped TiO₂/PVA composite solution was analyzed.

Results indicated that N-doped TiO₂ sample significantly increase absorbance wavelength toward visible light region than undoped TiO₂. Consequently, the band gap energy of the N-doped TiO₂ lower than the undoped TiO₂ sample. These features considerably increase the photo absorption property of TiO₂. Based on the analysis performed, both electrospun and doctor blade methods coated samples for power output of N-doped TiO₂ was higher than undoped TiO₂ which indicate, the chemical modification of TiO₂ by Nitrogen doping enhanced the photo catalytic efficiency of TiO₂. Furthermore, the power output of electrospun coated PV cell was higher than the doctor blade coated PV cell. Higher output of electrospun fiber coated PV cells attributed to the higher surface area and electron exchange ability of TiO₂ /PVA nanofibers. Therefore, electrospinning technique has successfully improved the photocatalytic efficiency of N-doped TiO₂ nanofibers.

ACKNOWLEDGEMENT

The authors would like to extend their sincerest gratitude to Mr Fareez Izwan Norman Wong for assistance during experiment and Universiti Teknologi MARA for funding the research from the internal grants as listed; 600-IRMI 5/3/GIP (041/2019), 600-IRMI/DANA 5/3/BESTARI (M) (016/2018), 600-IRMI/DANA 5/3/BESTARI (M) (017/2017), 600-IRMI/MyRA 5/3/BESTARI (021/2017).

REFERENCES

- X. Wu, X. Xu, C. Guo, and C. K. Ivan Tan, "Semiconductor Nanomaterials for Energy Conversion and Storage," *J. Nanomater.*, vol. 2015, pp. 1–2, 2015.
- I. Khan *et al.*, "Nanoparticles: Properties, applications and toxicities," *Arab. J. Chem.*, 2017.
- Y. Bai, I. Mora-Seró, F. De Angelis, J. Bisquert, and P. Wang, "Titanium Dioxide Nanomaterials for Photovoltaic Applications," *Chem. Rev.*, vol. 114, no. 19, pp. 10095–10130, 2014.
- J. Schneider *et al.*, "Understanding TiO₂ Photocatalysis: Mechanisms and Materials," *Chem. Rev.*, vol. 114, no. 19, pp. 9919–9986, 2014.
- A. Fujishima, X. Zhang, and D. A. Tryk, "TiO₂ photocatalysis and related surface phenomena," *Surf. Sci. Rep.*, vol. 63, no. 12, pp. 515–582, 2008.
- K. Prabarakan, S. Mohanty, and S. K. Nayak, "Improved electrochemical and photovoltaic performance of dye sensitized solar cells based on PEO/PVDF-HFP/silane modified TiO₂ electrolytes and MWCNT/Nafion[®] counter electrode," *RSC Adv.*, vol. 5, no. 51, pp. 40491–40504, 2015.
- P. Wang, L. Wang, B. Ma, B. Li, and Y. Qiu, "TiO₂ surface modification and characterization with nanosized PbS in dye-sensitized solar cells," *J. Phys. Chem. B*, vol. 110, no. 29, pp. 14406–14409, 2006.
- S. U. Halimi *et al.*, "Formation of Sol Gel Dried Droplets of Carbon Doped Titanium Dioxide (TiO₂) at Low Temperature via Electrospinning," *IOP Conf. Ser. Mater. Sci. Eng.*, vol. 358, no. 1, 2018.
- S. A. Bakar and C. Ribeiro, "Nitrogen-doped titanium dioxide: An overview of material design and dimensionality effect over modern applications," *J. Photochem. Photobiol. C Photochem. Rev.*, vol. 27, pp. 1–29, 2016.
- E. Katouezadeh, S. M. Zebardad, and K. Janghorban, "Synthesis and enhanced visible-light activity of N-doped TiO₂ nano-additives applied over cotton textiles," *J. Mater. Res. Technol.*, vol. 7, no. 3, pp. 204–211, 2018.
- S. M. Reda, M. Khairy, and M. A. Mousa, "Photocatalytic activity of nitrogen and copper doped TiO₂ nanoparticles prepared by microwave-assisted sol-gel process," *Arab. J. Chem.*, 2016.
- Y. M. Takayuki Doi *et al.*, "Single-Step Synthesis of Nanosized Titanium-Based Oxide/Carbon Nanotube Composites by Electro Spray Deposition and Their Electrochemical Properties," *J. Phys. Chem. C*, vol. 113, no. 18, pp. 7719–7722, 2009.
- D. Ha, Y. Yoon, and N. B. Zhitenev, "Nanoscale imaging of photocurrent enhancement by resonator array photovoltaic coatings," *Nanotechnology*, vol. 29, no. 14, 2018.
- H. Sun, J. Deng, L. Qiu, X. Fang, and H. Peng, "Recent progress in solar cells based on one-dimensional nanomaterials," *Energy Environ. Sci.*, vol. 8, no. 4, pp. 1139–1159, 2015.
- A. Haider, S. Haider, and I. K. Kang, "A comprehensive review summarizing the effect of electrospinning parameters and potential applications of nanofibers in biomedical and biotechnology," *Arab. J. Chem.*, vol. 11, no. 8, pp. 1165–1188, 2018.
- J. Xue, J. Xie, W. Liu, and Y. Xia, "Electrospun Nanofibers: New Concepts, Materials, and Applications," *Acc. Chem. Res.*, vol. 50, no. 8, pp. 1976–1987, 2017.
- S. Saallah, M. N. Naim, I. W. Lenggono, M. N. Mokhtar, N. F. Abu Bakar, and M. Gen, "Immobilisation of cyclodextrin glucanotransferase into polyvinyl alcohol (PVA) nanofibers via electrospinning," *Biotechnol. Reports*, vol. 10, pp. 44–48, 2016.
- I. Ismail, N. F. A. Bakar, T. H. Ling, N. Ideris, Z. H. M. Zain, N. Radaci, "Morphology and conductivity evaluation of electrospun polyacrylic acid (PAA) microfiber" *Mater. Today Proc.*, vol. 17, pp. 574–583, 2019.
- D. Li, G. Ouyang, J. T. McCann, and Y. Xia, "Collecting electrospun nanofibers with patterned electrodes," *Nano Lett.*, vol. 5, no. 5, pp. 913–916, 2005.
- A. S. Levitt, M. Alhabeb, C. B. Hatter, A. Sarycheva, G. Dion, and Y. Gogotsi, "Electrospun MXene/carbon nanofibers as supercapacitor electrodes," *J. Mater. Chem. A*, vol. 7, no. 1, pp. 269–277, 2019.
- P. Kallem, N. Yanar, and H. Choi, "Nanofiber-Based Proton Exchange Membranes: Development of Aligned Electrospun Nanofibers for Polymer Electrolyte Fuel Cell Applications," *ACS Sustain. Chem. Eng.*, vol. 7, no. 2, pp. 1808–1825, 2019.
- S. Chan *et al.*, "Electrospun carbon nanofiber catalyst layers for polymer electrolyte membrane fuel cells: Structure and performance," *J. Power Sources*, vol. 392, no. October 2017, pp. 239–250, 2018.
- W. A. Ayalew and D. W. Ayele, "Dye-sensitized solar cells using natural dye as light-harvesting materials extracted from Acanthus sennii chiovenda flower and Euphorbia cotinifolia leaf," *J. Sci. Adv. Mater. Devices*, vol. 1, no. 4, pp. 488–494, 2016.
- Z. Wu, F. Dong, W. Zhao, and S. Guo, "Visible light induced electron transfer process over nitrogen doped TiO₂ nanocrystals prepared by oxidation of titanium nitride," *J. Hazard. Mater.*, vol. 157, no. 1, pp. 57–63, 2008.
- G. Model, "Highly active TiO₂-x-yN_xF_y visible photocatalyst prepared under supercritical conditions in NH₄F/EtOH fluid," *Appl. Catal.*, vol. 203, pp. 1–8, 2009.
- C. Di Valentin *et al.*, "N-doped TiO₂: Theory and experiment," *Chem. Phys.*, vol. 339, no. 1–3, pp. 44–56, 2007.
- E. Pérez, M. F. Torres, G. Morales, V. Murgia, and E. Sham, "Synthesis of N-TiO₂ Effect of the Concentration of Nitrogen in the Band Gap," *Procedia Mater. Sci.*, vol. 8, pp. 649–655, 2015.
- M. Arvand *et al.*, "Fabrication of antibacterial silver nanoparticle-modified chitosan fibers using Eucalyptus extract as a reducing agent," *J. Appl. Polym. Sci.*, vol. 132, no. 25, pp. 1–8, 2015.
- D. I. Sanchez-Alvarado, J. Guzmán-Pantoja, U. Páramo-García, A. Maciel-Cerda, R. D. Martínez-Orozco, and R. Vera-Graziano, "Morphological study of chitosan/poly (vinyl alcohol) nanofibers prepared by electrospinning, collected on reticulated vitreous carbon," *Int. J. Mol. Sci.*, vol. 19, no. 6, pp. 1–12, 2018.
- K. Hari Prasad, S. Vinoth, P. Jena, M. Venkateswarlu, and N. Satyanarayana, "Structural characterization and impedance studies of PbO nanofibers synthesized by electrospinning technique," *Mater. Chem. Phys.*, vol. 194, no. June, pp. 188–197, 2017.
- D. H. Youn *et al.*, "Arbitrary alignment-angle control method of electrospun fibers: Potential for a stretchable electrode material," *RSC Adv.*, vol. 7, no. 71, pp. 44945–44953, 2017.

32. W. Qin, S. Lu, X. Wu, and S. Wang, "Dye-sensitized solar cell based on N-doped TiO₂ electrodes prepared on titanium," *Int. J. Electrochem. Sci.*, vol. 8, no. 6, pp. 7984–7990, 2013.
33. H. Melhem *et al.*, "Direct photocurrent generation from nitrogen doped TiO₂ electrodes in solid-state dye-sensitized solar cells: Towards optically-active metal oxides for photovoltaic applications," *Sol. Energy Mater. Sol. Cells*, vol. 117, pp. 624–631, 2013.
34. M. Fitra, I. Daut, M. Irwanto, N. Gomesh, and Y. M. Irwan, "Effect of TiO₂ thickness dye solar cell on charge generation," *Energy Procedia*, vol. 36, pp. 278–286, 2013.
35. S. Nagata, G. M. Atkinson, D. Pestov, G. C. Tepper, and J. T. M. Jr, "Electrospun Polymer-Fiber Solar Cell," vol. 2013, 2013.



Assoc. Professor Dr. Mohd. Nazli Naim Graduated in PhD. in Engineering, Tokyo Univ. of Agriculture and Tech. (TUAT), Japan (東京農工大学), His 27 articles were published and indexed in (Scopus). His Research work in Nanobiotechnology since 2010. He is a Chartered Engineer (C.Eng) from Institution of Chemical Engineers (ICHEM), U. Kingdom. He obtained 10 medals and awards in International/local symposium.

AUTHORS PROFILE



Mohamed Hasaan Hussain received his BEng (Hons) in Mechanical engineering from University of Sunderland, UK in 2015. He is currently pursuing his MSc at the department of Chemical engineering, University Teknologi MARA (UiTM), Malaysia under guidance of Associate Professor Dr. Noor Fitrah Abu Bakar. His current research interests focus on metal nanoparticle synthesis and, electrospray and electrospinning technology incorporated nanomaterial preparation.



Assoc. Professor Dr. Noor Fitrah Abu Bakar (corresponding author) received her PhD. in Engineering from Tokyo Univ. of Agriculture and Tech. (TUAT), Japan. She is a Chartered Engineer (C.Eng., IChemE, UK). She is a senior lecturer in Faculty of Chemical Engineering, Universiti Teknologi MARA, Shah Alam, Selangor. She published 38 indexed articles related to particles processing, electric force assisted technique in nanotechnology and synthesis of nanomaterials using microwave technology. Her current SCOPUS H-index is 7 with 82 citations.



Dr. Lim Ying Pei received her bachelor's degree in chemical engineering from Universiti Kebangsaan Malaysia (2001), MEng. in Environmental (2003) and PhD. in Chemical Engineering from the same university in 2017. She joined Sepakat Setia Perunding (M) Sdn Bhd as consulting engineer in 2003. Since 2005, she has been with the Faculty of Chemical Engineering, Universiti Teknologi MARA, where she is currently senior lecturer. Her main areas of research interest are membrane separation technology, photocatalytic degradation and wastewater treatment. She has received several research grants in the area of ultrafiltration and photocatalysis. Dr Lim has published more than 22 academic papers in referred journal, book chapters and proceedings. Her h-index is 9 with 342 citations in Google Scholar Citation. Dr. Lim is an associate member of the Institute of Chemical Engineers of Malaysia and member for Board of Engineers Malaysia and MyMembrane



Dr. Lim Ying Chin received her bachelor's degree in chemistry with Education from Universiti Teknologi Malaysia (2001), followed by a MSc. in Chemistry from the same university in 2002. She then received her Ph.D in Material Science from Universiti Putra Malaysia in 2013. Dr. Lim is currently a senior lecturer at Faculty of Applied Sciences, Universiti Teknologi MARA, Malaysia. Her major research involves fabrication and characterization of titania nanotubes for photoelectrochemical application. Her research interests include self-ordering porous oxide nanostructures on valve metal via electrochemical route for solar cell application and photocatalysis. She has received several government/university research grants in the areas of materials chemistry, acting as the head researcher or associate researcher for such grants. She has also received international Toray Science & Technology Research Grant funded by Toray Science Foundation, Japan. Dr. Lim is also a member of American Chemical Society (ACS) and Malaysian Institute of Chemistry (IKM). Dr. Lim has published several academic writings related to her field in journals (30 refereed journals), and chapters in books (3) and the rest is conference proceedings (9). Her H-index is 5 with 94 citations (Scopus).