

Planning and Characterization of Green Synthesized Ferric Oxide (Fe₂O₃) Nanoparticles

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ABSTRACT--The ongoing improvement and execution of new advancements have prompted new time, the nano-transformation which unfurls the job of plants in bio and green combination of nanoparticles which appear to have drawn significant unequivocal consideration from a perspective of blending stable nanoparticles. Green standard courses of orchestrating have risen as a choice to defeat the confinement of ordinary strategies among which plants and microorganisms are significantly misused. Thus the present investigation imagines the biosynthesis of nanoparticles from plants which are developing as nanofactories. Ferric Oxide (Fe₂O₃) nanoparticles were incorporated via completion of water under the surrounding conditions. Ferrous Sulfate and Sulfur were broken up in Salt Petra with a molar proportion of 2: 1. Citrate particles were utilized as nucleation stabilizers. The reactor was loaded up with a latent argon environment and the NPs were gradually accelerated by dropping of KAI (SO₄) under an overwhelming blend. The moderate arrangement of nanoparticle seeds was trailed by a quicker development of centers and a moderate development of shells balanced out by citrate particles for the entire time. The tanish red item was accelerated with CH₃CO, centrifuged for 5 minutes at 2,500 rpm, and the pellet scattered in argon-foamed water. This means they were rehashed twice to dependably wash the NPs. In the present work the writer contemplates Synthesis and Characterization of Fe₂O₃ nanoparticles.

Index Terms—Plants, Nanoparticles, Green synthesis, Ferrous Sulphate, Fe₂O₃ and SEM

I. INTRODUCTION

The rise of nanotechnology has given a broad research as of late by crossing with different parts of science and shaping effect on all types of life. The idea of nanotechnology was initially started with an address conveyed by Richard Feynman in 1959. Nanotechnology is a field of science which manages generation, control and utilization of materials extending in nanometers. In nanotechnology nanoparticles inquire about is an imperative viewpoint because of its incalculable applications. Nanoparticles have communicated huge advances attributable to the wide scope of utilizations in the field of Manufacturing, bio-medical, sensors, antimicrobials, impetuses, hardware, optical filaments, horticultural, bio-marking and in different zones.

II. FABRICATION OF NANOPARTICLES FROM PLANTS

The union of sporadic state of Ferrous nanoparticles from the extracellular fluid dried clove buds (Syzygium aromaticum) was accounted for and FTIR portrayal uncovered that the unreservedly water solvent flavonoids of clove buds are in charge of bioreduction of Ferrous particles. Essentially, unrefined ethyl-acetic acid derivation concentrate of Ulvafasciata, was assessed for nanoparticles union brought about polydispersed nanoparticles with size extending from 28-41nm. The combination of antimicrobial ferrous nanoparticles utilizing tissue culture-determined callus and leaf of the saltmarsh plant, Sesuvium portulacastrum L. was considered. The callus separate could deliver ferrous nanoparticles, superior to anything leaf extricate. The combination was affirmed by utilizing X-beam diffraction range. TEM brought about the arrangement of Ferrous nanoparticles with a round formed and the size going from 5-20nm. Fourier change infrared (FTIR) spectroscopy uncovered the nearness of proteins, flavones and terpenoids which were in charge of the adjustment of the ferrous nanoparticles. The orchestrated iron nanoparticles indicated huge action against clinical strains of microscopic organisms than the fungi¹⁵. The fluid concentrate of flower petals was utilized for the investigation of biosynthesis of gold nanoparticles showed gold nanoparticles upon described by UV- VIS spectroscopy, FT-IR spectroscopy, X-beam diffraction and vitality dispersive X-beam spectroscopy. FT-IR spectroscopy uncovered the nearness of biomolecules that have essential amine gathering (– NH₂), carbonyl gathering, – OH gatherings and other balancing out useful gatherings that are in charge of the adjustment of gold nanoparticles. X-beam diffraction design indicated high virtue and face focused cubic structure of gold nanoparticles. The measure of gold nanoparticles was dictated by Dynamic light dissipating system and it was observed to be around 10 nm¹⁸. The bio creation of gold nanoparticles came about to rely upon various parameters like temperature and pH impacts on its combination utilizing the watery concentrate of Macrotyloma uniflorum. Biosynthesized nanoparticles were recorded by UV- noticeable spectroscopy, transmission electron microscopy (TEM), X-beam diffraction (XRD) and FTIR investigation. The high crystallinity with FCC period of nanoparticles was broke down by HRTEM pictures, SAED and XRD designs. The extent of the nanoparticles was extending from 14-17nm and FTIR

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range came about the nearness of various practical gatherings present in the bio-particle topping the nanoparticles.

A.Applications of plant mediated synthesized Nanoparticles

Bio-manufactured nanoparticles utilized for Biological Assays: Rapid orchestrate of ferrous nanoparticles utilizing leaf concentrate of *Acalypha indica* was assessed against the water borne bacterial pathogens. Silver nanoparticles portrayal was recorded from UV– Vis range, checking electron microscopy (SEM), X-beam diffraction (XRD) and vitality dispersive spectroscopy (EDS) and brought about the arrangement of 20-30nm molecule size of nanoparticles. The antibacterial action of incorporated Copper nanoparticles indicated successful inhibitory movement against *Escherichia coli* and *Vibrio cholera* with MIC extending 10µg/ml. Ferrous nanoparticles from Ferric sulfate with Sulfur as a decreasing specialist in nearness of potassium Nitric with alum. The ferrous nanoparticles utilizing plant leaf concentrate of *Magnolia*. An Electron microscopy investigation uncovered a measure of ferrous nanoparticles of around 40 to 100 nm upon an assessment brought about by the antibacterial movement against *Escherichia coli* and watched noteworthy action against the test pathogen. Ferrous nanoparticles were incorporated by utilizing papaya natural product as decreasing just as topping specialist. The creation of nanoparticles was observed by utilizing UV– Vis retention spectroscopy and was portrayed by FTIR, XRD and SEM. The X-beam diffraction and SEM investigation demonstrated the normal molecule size of 15 nm just as uncovered their cubic structure. Combined nanoparticles were assessed for antimicrobial action against multi-medicate safe human pathogens. Antimicrobial action of silver nanoparticles blended with *Psidium guajava* was assessed against human pathogens. The orchestrated silver nanoparticles indicated a huge antimicrobial movement against *Escherichia coli*, *Bacillus cereus* and *Candida tropicalis*. Ferrous nanoparticles by utilizing the rhizome concentrate of *Dioscorea batatas* and were portrayed by UV-Vis spectrophotometer, SEM, FTIR, XRD, and EDX. Upon antimicrobial assessment came movement against gram positive (*B. subtilis* and *S. aureus*), gram negative (*E. coli*), and organisms (*S. cerevisiae* and *C. albicans*). The blend of ferrous nanoparticles utilizing the *Cassia auriculata* leaf extricate was and assessed for antimicrobial action against *E.coli*, *Serratiamarcescens*, *Bacillus subtilis*, *Aspergillusniger* and *Aspergillusflavus*. Parasites indicated critical action pursued by the test microscopic organisms. Biosynthesized ferrous nanoparticles utilizing stem bark concentrates of *Boswellia* and *Shorea*; and leaf concentrate of *Svensonia* was assessed against the board of pathogenic microorganism viz., *Proteus*, *Pseudomonas*, *Klebsiella*, *Bacillus* and *E.coli* types of microscopic organisms and *Aspergillus*, *Fusarium*, *Curvularia* and *Rhizis* types of growths. Ferrous nanoparticles blended from bark concentrates of *Boswellia ovalifoliolata* and *Shorea tumbuggaia* demonstrated critical movement against

Klebsiella and *Aspergillus*; and *Pseudomonas* and *Fusarium* species individually though the leaf concentration of *Svensonia hydrobadensis* hindered the development of *Pseudomonas* and *Rhizopus* species. The development of ferrous nanoparticles utilizing stem concentration of *Svensonia hydrobadensis* (Walp.) Mold. were tried against microscopic organisms, for example, *Proteus*, *Pseudomonas*, *Klebsiella*, *Bacillus* and *E.coli* and growths *Aspergillus*, *Fusarium*, *Curvularia* and *Rhizopus* species. The antimicrobial movement of silver nanoparticles demonstrated huge action against *Pseudomonas*, *Curvularia* and *Fusarium* and moderate action against *E. coli*, *Klebsiella*, *Bacillus*, *Proteus*, *Aspergillus flavus* and *Aspergillus niger*.

B. Nanoparticles Synthesis

Varieties of ordinary strategies have been utilized in combination of nanoparticles. In any case, these customary techniques are bound with different constraints, for example, costly, age of unsafe poisonous synthetic substances and so on., which has upsurge the scientists to create safe, eco-accommodating elective methodologies in combination of nanoparticles among which organic frameworks have been engaged and misused as a favored green standard process for amalgamation of nanoparticles. Without a doubt, natural frameworks have a one of a kind capacity for generation of exact shape and controlled structures. The current survey accentuates revealed plant assets for the union of various nanoparticles. Plants are known to have different restorative mixes which have been abused since old times as a conventional work. Due to its enormous assorted variety of plants have been investigated continually for wide scope of uses in the field of Production, Design and modern and so on. Late reports of plants towards generation of nanoparticles are said to have focal points, for example, effectively accessible, safe to deal with and wide scope of biomolecules, for example, alkaloids, terpenoids, phenols, flavanoids, tannins, quinines and so on are known to intervene union of nanoparticles. The spread of antimicrobial opposition expands the odds of significant harms for development of scraping, setting and so forth., on metals. Accordingly finding new and compelling answers for keep these types of harms, the cutting edge nanotechnology has ended up being a successful countermeasure to handle the danger of diseases. On this note, later logical leaps forward have exhibited that antimicrobial nanomaterials are successful in keeping infective specialists from creating opposition. As of late, science has investigated increasingly complex antimicrobial coatings and nonmaterial dependent on Copper and Iron, which have appeared potential in antibacterial treatment. The reason for this article is to develop the talk on the danger of contamination identified with surface purification, and to survey the cutting edge and potential arrangements, with explicit spotlight on sanitization strategies utilizing nanomaterials.

C. Properties of Nanomaterials

Nanostructured materials include single stage or multiphase polycrystalline solids with a run of the mill the normal size of a couple

nanometers(1nm=10⁻⁹m).Basically, the range from (1-100nm)is taken as nano-go for tradition according to the National Nanotechnology Initiative in the US, and the measure of hydrogen iota is considered as the lower furthest reaches of nano while maximum limit is

subjective. The grain sizes are so small: a critical volume division of the molecules dwells on grain limits. The material is described by an expansive number of interfaces in which the nuclear course of action is not quite the same as those of gem grid. The fundamental characterization of nonmaterials is dependent on repression. Mass structures demonstrate no imprisonment, though nano-wells and non-wires can be obtained by 2-D and 1-D repression and prompts zero measurement quantum structures that are quantum specks.

D.Mechanical Properties

Because of the nanometer estimate, a large number of the mechanical properties of the nanomaterials are not the same as the mass materials including the hardness, flexible modulus, break sturdiness, scratch obstruction and weariness quality and so forth., an upgrade of mechanical properties of Nanomaterials can result because of this change, which are for the most part coming about because of basic flawlessness of the materials. The small size either renders them free of inner auxiliary blemishes, for example, disengagements, small scale twins, and pollution encourages or the few deformities or debasements present can't duplicate adequately to cause mechanical disappointment. The defects inside the nano measurement are very enthusiastic and will relocate to the surface to loosen up themselves under strengthening, decontaminating the material and leaving ideal material structures inside the Nanomaterials. In addition, the outside surfaces of Nanomaterials additionally have less or free of deformities contrasted with mass materials, serving to improve the mechanical properties of Nanomaterials. The upgraded mechanical properties of the Nanomaterials could have numerous potential applications both in nanoscale, for example, mechanical nano resonators, sensors, magnifying instrument test tips and nanotweezers for nano scale object control, light weight high quality materials, adaptable conductive coatings, wear opposition coatings, harder and harder cutting apparatuses and so forth.,

innovation, mechanical cleaning and so on., the fundamental reason for change in various mechanical,thermal and other property is because of increment in surface to volume proportion.

B.Fabrication of Ferric Oxide by Bio synthesis method

Iron Oxide and Ferric Oxide (Fe₂O₃) nanoparticles were integrated via did in water under encompassing conditions. Ferrous Sulfate (Fig 1) and Sulphur (Fig 2) were broken down in Salt Petra with a molar proportion of 2: 1. Citrate ions (Fig 7) were utilized as nucleation stabilizers. The reactor was loaded up with inactive argon air and the NPs were gradually accelerated by dropping of KAI (SO₄)(Fig 4 and 5) under overwhelming blending. The moderate development of nanoparticle seeds was trailed by a quicker arrangement of centers and a moderate arrangement of shells balanced out by citrate particles for the entire time. The caramel red item was accelerated with CH₃)₂CO, centrifuged for 5 minutes at 2,500rpm, and the pellet scattered in argon-percolated water. This means they were rehashed twice to dependably wash the NPs.

C. Fe₂O₃ Nano particles

It normally shows up as a metallic powder (Fig 8) and is almost insoluble in water. The powder is broadly utilized as an added substance for various materials and items including plastics, earthenware production, glass, bond, elastic (for example vehicle tires), greases, paints, treatments, glues, sealants, colors, sustenances, batteries, ferrites, fire retardants, and so forth. Fe₂O₃ is available on Earth covering as a mineral center iron in any case; most Fe₂O₃ utilized financially is delivered artificially. Fe₂O₃ is nontoxic and is good with human skin, making it a reasonably added substance for materials and surfaces that interact with the human



Figure 1: Ferrous Sulphate



Figure 2: Sulphur

III. STEPS INVOLVED IN THE BIOSYNTHESIS OF NANOPARTICLES

A.Processing Methods

The combination of Nanomaterial can be very much achieved by two methodologies. Right off the bat, with a "Base Up" strategy where little building squares are delivered and collected into bigger structures. Where the principle controls parameters is morphology, crystallinity, molecular size, and synthetic piece. Precedents: concoction combination, laser catching, self-gathering, colloidal total, and so forth., and also, by "Top Down" strategy where huge items are altered to give littler highlights. For instance: film statement and growth,nano engrave/lithography,etching



Figure 3: Alum



Figure 4: Potassium Nitric



Fig 5: Bonding of potassium nitric



Fig6: Cow dung cake



Fig 7: citrus acid



Fig 8: Fe₂O₃ Nanoparticles synthesized through Bio synthesis method

IV. RESULTS & DISCUSSIONS

The surface morphology, homogeneity and grain size of the deposited films were studied by Scanning Electron Microscopy (SEM).

A. Morphological analysis of Fe₂O₃ Nanoparticles for the effect of annealing temperatures

Surface morphology of Fe₂O₃ nanoparticles is acquired at various bar scale with various magnification at 3µm at 2.0kV appeared in Figure 9.0. It is noticed that the surface is found to uniform and secured with unshaped grains for molecule acquired at 1µm at 5.0K and 10.0k (Figure 9.1 and 9.2). Figure 9.3 demonstrates that the surface is seen to smooth with inclusion of hexagonal formed grain for molecule acquired at 1µm at 10.0k Magnification, the Nanoparticles size to be in the middle of 258.8nm-301.4nm. In this manner the SEM considers showing that smooth uniform surface with uniform grain dispersion is acquired for the Fe₂O₃ nanoparticles size to be 258.8nm-301.4nm.

B. Morphological analysis of Fe₂O₃ nanoparticles for the effect of Molarity

The surface morphology of Fe₂O₃ nanoparticles arranged with 0.2M, 0.3M and 0.4M of ferric acetic acid derivation was completed utilizing Scanning Electron Microscope (SEM). The surface morphology of the Fe₂O₃ nanoparticles at various fixations was portrayed in Figure 9.3. All of the Fe₂O₃ films have granular and uniform grains at the request of nm. It very well may be seen that the grain size of Fe₂O₃ film increments with increment in ferric acetic acid derivation focus. As the focus builds on, the gem grain grows ceaselessly. The SEM thinks about showing that a smooth uniform surface with uniform grain appropriation is received for the Fe₂O₃ nanoparticles arranged with an upgraded 0.3 Molarity focus.

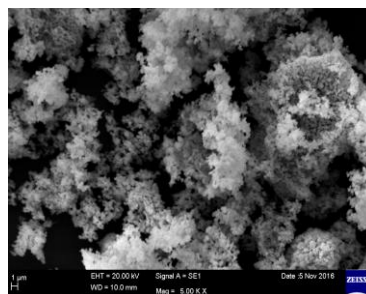


Figure 9. Fe₂O₃ Barscale 3µm at 2.0kMag

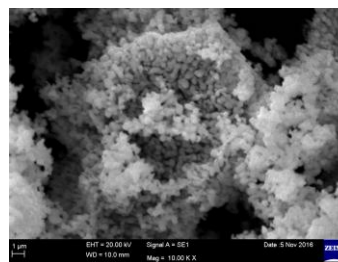


Figure 10. Fe₂O₃ Barscale 1µm at 5.0kMag

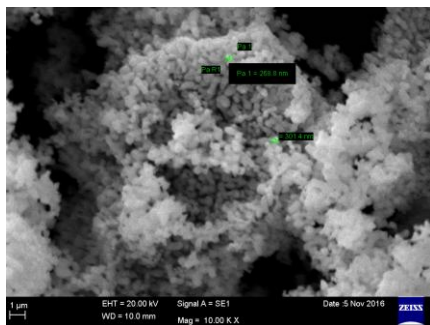


Figure 11. Fe₂O₃ Barscale 1μm at 10.0kMag(258.8nm-301nm)

V. CONCLUSIONS

An organic blend of nanoparticles has an upsurge in the field of nano-biotechnology to make novel materials that are eco-accommodating, practical, stable nanoparticles with an incredible significance for more extensive applications in the territories of gadgets, medication, horticulture and mechanical field. Amid the present situation nanotechnology persuades advance in all circles of life, henceforth biosynthetic course of nanoparticles amalgamation will develop as more secure and best option in contrast to ordinary techniques. In spite of the fact that different natural substances have been abused for the creation of nanoparticles, the utilization of plants for the easy hearty combination of nanoparticles is enormous. In this manner the present audit imagines the significance of plant interceded nanoparticles creations by giving the different written works revealed by a wide margin. With tremendous plant assorted varieties considerably more plant species are in approach to being abused and announced in future time towards fast and single step convention with green guidelines. The SEM examines shows that a smooth uniform surface with uniform grain dissemination is acquired for the Fe₂O₃ nanoparticles arranged with an enhanced 0.3 Molarity focus.

REFERENCES

1. H. B. Na, I. C. Song, and T. Hyeon, "Inorganic nanoparticles for MRI contrast agents," *Advanced Materials*, 21, 21, 2133–2148, 2009.
2. J. Y. Park, M. J. Baek, E. S. Choi et al., "Paramagnetic ultrasmall gadolinium oxide nanoparticles as advanced T1 MRI contrast agent: account for large longitudinal relaxivity, optimal particle diameter, and in Vivo T1 MR Images," *ACS Nano*, 3, 11, 3663–3669, 2009.
3. F. Hu and Y. S. Zhao, "Inorganic nanoparticle-based T1 and T1/T2 magnetic resonance contrast probes," *Nan scale*, 4, 20, 6235–6243, 2012.
4. F. Hu, K. W. MacRenaris, E. A. Waters et al., "Ultrasmall, water-soluble magnetite nanoparticles with high relaxivity for magnetic resonance imaging," *The Journal of Physical Chemistry C*, 113, 49, 20855–20860, 2009.
5. M. Rohrer, H. Bauer, J. Mintorovitch, M. Requardt, and H.-J. Weinmann, "Comparison of magnetic properties of MRI contrast media solutions at different magnetic field strengths," *Investigative Radiology*, 40, 11, 715–724, 2005.
6. D. Frascione, C. Diwoy, G. Almer et al., "Ultrasmall superparamagnetic iron oxide (USPIO)-based liposomes as magnetic resonance imaging probes," *International Journal of Nanomedicine*, 7, 2349–2359, 2012.

7. R. Qiao, C. Yang, and M. Gao, "Superparamagnetic iron oxide nanoparticles: from preparations to in vivo MRI applications," *Journal of Materials Chemistry*, 19, 35, 6274–6293, 2009.
8. C. F. G. C. Geraldes and S. Laurent, "Classification and basic properties of contrast agents for magnetic resonance imaging," *Contrast Media and Molecular Imaging*, 4, 1, 1–23, 2009. A. J. L. Villaraza, A. Bumb, and M. W. Brechbiel, "Macromolecules, dendrimers, and nanomaterials in magnetic resonance imaging: the interplay between size, function, and pharmacokinetics," *Chemical Reviews*, 110, 5, 2921–2959, 2010.
9. G. Huang, H. Chen, Y. Dong et al., "Superparamagnetic iron oxide nanoparticles: amplifying ros stress to improve anticancer drug efficacy," *Theranostics*, 3, 2, 116–126, 2013.
10. C. Jiang, J. Jia, and S. Zhai, "Mechanistic understanding of toxicity from nanocatalysts," *International Journal of Molecular Sciences*, 15, 8, 13967–13992, 2014.
11. L. Tong, M. Zhao, S. Zhu, and J. Chen, "Synthesis and application of superparamagnetic iron oxide nanoparticles in targeted therapy and imaging of cancer," *Frontiers of Medicine*, 5, 4, 379–387, 2011.
12. T. Sugimoto and E. Matijević, "Formation of uniform spherical magnetite particles by crystallization from ferrous hydroxide gels," *Journal of Colloid and Interface Science*, 74, 1, 227–243, 1980.
13. R. Massart, "Preparation of aqueous magnetic liquids in alkaline and acidic media," *IEEE Transactions on Magnetics*, 17, 2, 1247–1248, 1981.
14. A. Bee, R. Massart, and S. Neveu, "Synthesis of very fine maghemite particles," *Journal of Magnetism and Magnetic Materials*, 149, 1-2, 6–9, 1995.
15. S. Laurent, D. Forge, M. Port et al., "Magnetic iron oxide nanoparticles: synthesis, stabilization, vectorization, physicochemical characterizations, and biological applications," *Chemical Reviews*, 108, 6, 2064–2110, 2008.
16. M. Safdarian, P. Hashemi, and M. Adeli, "One-step synthesis of agarose coated magnetic nanoparticles and their application in the solid phase extraction of Pd(II) using a new magnetic field agitation device," *Analytica Chimica Acta*, 774, 44–50, 2013.
17. C. Tassa, S. Y. Shaw, and R. Weissleder, "Dextran-coated iron oxide nanoparticles: a versatile platform for targeted molecular imaging, molecular diagnostics, and therapy," *Accounts of Chemical Research*, 44, 10, 842–852, 2011.
18. R. G. López, M. G. Pineda, G. Hurtado et al., "Chitosan-coated magnetic nanoparticles prepared in one step by reverse microemulsion precipitation," *International Journal of Molecular Sciences*, 14, 10, 19636–19650, 2013.
19. H. Yilmaz and S. H. Sanlier, "Preparation of magnetic gelatin nanoparticles and investigating the possible use as chemotherapeutic agent," *Artificial Cells, Nanomedicine and Biotechnology*, 41, 2, 69–77, 2013.
20. M. M. Yallapu, S. P. Foy, T. K. Jain, and V. Labhasetwar, "PEG-functionalized magnetic nanoparticles for drug delivery and magnetic resonance imaging applications," *Pharmaceutical Research*, 27, 11, 2283–2295, 2010.
21. A. Petri-Fink, B. Steitz, A. Finka, J. Salaklang, and H. Hofmann, "Effect of cell media on polymer coated super paramagnetic iron oxide nanoparticles (SPIONs): colloidal stability, cytotoxicity, and cellular uptake studies," *European Journal of Pharmaceutics and Biopharmaceutics*, 68, 1, 129–137, 2008.

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22. L. P. Ramírez and K. Landfester, "Magnetic polystyrene nanoparticles with a high magnetite content obtained by miniemulsion processes," *Macromolecular Chemistry and Physics*, 204, 1, 22–31, 2003.
23. T. Linemann, L. B. Thomsen, K. G. Du Jardin et al., "Development of a novel lipophilic, magnetic nanoparticle for in vivo drug delivery," *Pharmaceutics*, 5, 2, 246–260, 2013.
24. S. Gyergyek, M. Drogenik, and D. Makovec, "Oleic-acid-coated CoFe₂O₄ nanoparticles synthesized by co-precipitation and hydrothermal synthesis," *Materials Chemistry and Physics*, 133, 1, 515–522, 2012.
25. M. Iijima, Y. Yonemochi, M. Kimata, M. Hasegawa, M. Tsukada, and H. Kamiya, "Preparation of agglomeration-free hematite particles coated with silica and their reduction behavior in hydrogen," *Journal of Colloid and Interface Science*, 287, 2, 526–533, 2005.
26. M. Kosmulski, "Positive electrokinetic charge of silica in the presence of chlorides," *Journal of Colloid and Interface Science*, 208, 2, 543–545, 1998.
27. R. Xu, *Particle Characterization: Light Scattering Methods*, Springer, Dordrecht, The Netherlands, 2006.
28. A. Hlavacek and P. Skládal, "Isotachophoretic purification of nanoparticles: tuning optical properties of quantum dots," *Electrophoresis*, 33, 9-10, 1427–1430, 2012.
29. J. Hennig, A. Nauerth, and H. Friedburg, "RARE imaging: a fast imaging method for clinical MR," *Magnetic Resonance in Medicine*, 3, 6, 823–833, 1986.
30. C. M. Hoo, N. Starostin, P. West, and M. L. Mecartney, "A comparison of atomic force microscopy (AFM) and dynamic light scattering (DLS) methods to characterize nanoparticle size distributions," *Journal of Nanoparticle Research*, 10, 1, 89–96, 2008.